

Copper-Catalyzed One-Pot Synthesis of *N*-Sulfonylalkanimidoyl Thiocyanates from Sulfonyl Azides, Alkynes, and KSCN

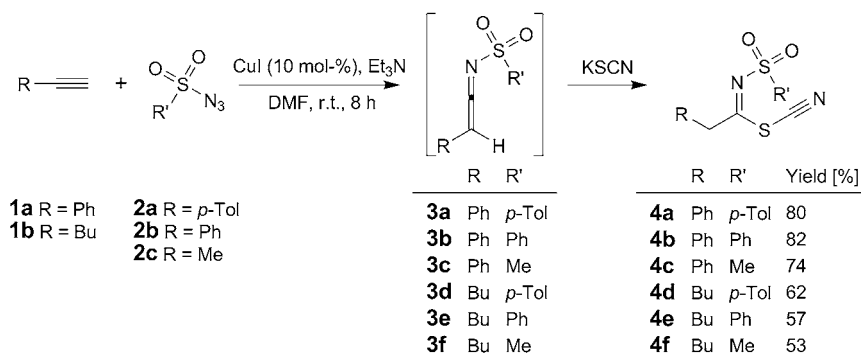
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Ketenimine intermediates generated by the addition of copper acetylides to sulfonyl azides are trapped by KSCN to afford *N*-sulfonylalkanimidoyl thiocyanates in moderate-to-good yields.

Introduction. – Ketenimines [1], as useful intermediates, have attracted much attention due to their diverse chemistry and relative reactivity [2–5]. Perhaps, the most attractive method to generate ketenimines is the Cu^I-catalyzed azide–alkyne cycloaddition (*Scheme 1*), which was established by Sharpless and co-workers [6]. This method is well suited for multicomponent reactions [7]. The formation of ketenimine intermediates from terminal alkynes, sulfonyl azides, and Et₃N as the base, in the presence of CuI [8], has encouraged us to trap these intermediates using nucleophilic addition reaction [9–11].

Scheme 1



Herein, we report a simple and efficient procedure for the synthesis of *N*-sulfonylalkanimidoyl thiocyanates *via* the Cu-catalyzed three-component coupling reaction of sulfonyl azides, terminal alkynes, and KSCN (*Scheme 1*).

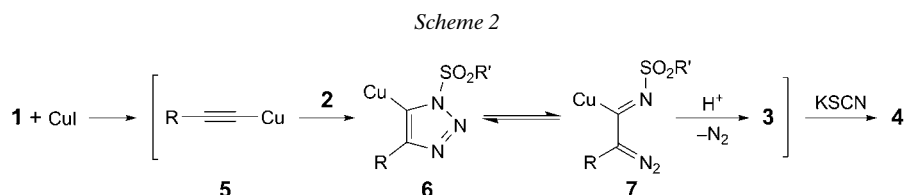
Results and Discussion. – In our initial investigations, phenylacetylene (**1a**), *p*-toluenesulfonyl azide (TsN₃; **2a**), and KSCN were selected as the model substrates. Several catalysts such as CuI, CuBr, CuCl, Cu₂O, and Cu powder were tested, with CuI giving the best results. Among several solvents screened, DMF was the best. When the

reaction was performed in DMF in the presence of 1 equiv. of Et₃N at room temperature for 8 h, *N*-sulfonyl-2-phenylethanimidoyl thiocyanate (**4a**) was obtained in 80% yield. Thus, the optimized reaction conditions used were 10 mol-% of CuI, 1 mmol of alkyne, 1.2 mmol of sulfonyl azide, and 1.5 mmol of KSCN in 3 ml of DMF at room temperature.

Aliphatic acetylenes were as low-yielding substrates compared to phenylacetylene. Aromatic and aliphatic sulfonyl azides reacted efficiently, and the corresponding products were obtained in good yields.

Structures of compounds **4a–4f** were assigned by IR, ¹H- and ¹³C-NMR, and mass-spectral data. The ¹H-NMR spectrum of **4a** exhibited two *singlets* for Me (δ (H) 2.48) and CH₂ (δ (H) 3.69) H-atoms, along with characteristic *multiplets* for the aromatic H-atoms. The ¹³C-NMR spectrum of **4a** displayed twelve signals in agreement with the proposed structure. The mass spectrum of **4a** showed the molecular-ion peak at *m/z* 330. The NMR spectra of compounds **4b–4f** were similar to those of **4a**, except for the substituents, which gave rise to characteristic signals in the appropriate regions of the spectra.

A plausible mechanism for the formation of compounds **4** is proposed in *Scheme 2*. The yellow copper acetylide **5**, formed from **1** and CuI, is converted to ketenimine **3** by well-documented transformations [12–14]. Nucleophilic attack of the thiocyanate ion at **3** affords **4**.



In summary, ketenimine intermediates generated by the addition of copper acetylides to sulfonyl azides are trapped by KSCN to yield *N*-sulfonylalkanimidoyl thiocyanates. The potential diversity of this type of reaction, and available starting materials and catalysts are the main advantages of this methodology.

Experimental Part

General. All chemicals were obtained commercially and used without further purification. IR Spectra: *Shimadzu-IR-460* spectrometer; $\tilde{\nu}$ in cm⁻¹. ¹H- and ¹³C-NMR Spectra: *Bruker DRX-500 Avance* instrument at 500.1 and 125.7 MHz, resp.; δ in ppm, *J* in Hz. MS: *Finnigan-MAT-8430EI-MS* mass spectrometer; at 70 eV; in *m/z* (rel. %). Elemental analyses: *Vario EL III CHNOS* elemental analyzer.

General Procedure for the Synthesis of Compounds 4. To a mixture of alkyne **1** (1 mmol), sulfonyl azide **2** (1.2 mmol), CuI (0.1 mmol), and Et₃N (1 mmol) in DMF (3 ml) was slowly added KSCN (1.5 mmol). The mixture was stirred at r.t. After completion of the reaction (*ca.* 8 h; TLC (AcOEt/hexane 1:3) monitoring), the mixture was diluted with CH₂Cl₂ (2 ml) and aq. NH₄Cl soln. (3 ml), stirred for 30 min, and the layers were separated. The aq. layer was extracted with CH₂Cl₂ (3 × 3 ml), and the combined org. fractions were dried (Na₂SO₄) and concentrated under reduced pressure. The residue was purified by flash column chromatography (SiO₂, 230–400 mesh, *Merck*; AcOEt/hexane 1:3) to give the product.

N-[(4-Methylphenyl)sulfonyl]-2-phenylethanimidoyl Thiocyanate (**4a**). Yellow oil. Yield: 0.26 g (80%). IR (KBr): 2253, 1511, 1396, 1270, 1139, 1083. ¹H-NMR: 2.48 (s, Me); 3.69 (s, CH₂); 7.26–7.34 (m, 3 arom. H); 7.40 (d, ³J=7.6, 2 arom. H); 7.50 (d, ³J=7.9, 2 arom. H); 7.92 (d, ³J=7.9, 2 arom. H). ¹³C-NMR: 32.2 (Me); 42.2 (CH₂); 115.1 (CN); 125.7 (CH); 127.5 (2 CH); 128.8 (2 CH); 129.2 (2 CH); 130.7 (2 CH); 132.5 (C); 142.1 (C); 147.3 (C); 170.3 (C). MS: 330 (2, M⁺), 272 (8), 238 (12), 155 (100), 91 (70), 77 (54), 57 (35). Anal. calc. for C₁₆H₁₄N₂O₂S₂ (330.42): C 58.16, H 4.27, N 8.48; found: C 58.47, H 4.34, N 8.59.

2-Phenyl-N-(phenylsulfonyl)ethanimidoyl Thiocyanate (**4b**). Yellow oil. Yield: 0.26 g (82%). IR (KBr): 2249, 1520, 1400, 1385, 1273, 1142, 1087. ¹H-NMR: 3.67 (s, CH₂); 7.25–7.34 (m, 3 arom. H); 7.48 (d, ³J=7.6, 2 arom. H); 7.62 (t, ³J=7.6, 1 arom. H); 7.76 (t, ³J=7.9, 2 arom. H); 8.05 (d, ³J=7.9, 2 arom. H). ¹³C-NMR: 42.5 (CH₂); 115.1 (CN); 125.6 (CH); 127.4 (2 CH); 128.7 (2 CH); 129.2 (2 CH); 130.1 (2 CH); 132.5 (CH); 135.6 (C); 144.9 (C); 170.9 (C). MS: 316 (3, M⁺), 258 (12), 238 (17), 175 (31), 141 (100), 91 (45), 77 (54), 57 (46). Anal. calc. for C₁₅H₁₂N₂O₂S₂ (316.40): C 56.94, H 3.82, N 8.85; found: C 57.36, H 3.89, N 8.96.

N-(Methylsulfonyl)-2-phenylethanimidoyl Thiocyanate (**4c**). Yellow oil. Yield: 0.19 g (74%). IR (KBr): 2245, 1531, 1401, 1385, 1271, 1120. ¹H-NMR: 3.44 (s, Me); 3.64 (s, CH₂); 7.25–7.36 (m, 3 arom. H); 7.49 (d, ³J=7.6, 2 arom. H). ¹³C-NMR: 33.0 (Me); 42.5 (CH₂); 115.4 (CN); 125.6 (C); 128.8 (2 CH); 129.3 (2 CH); 132.5 (CH); 170.3 (C). MS: 254 (2, M⁺), 196 (10), 175 (14), 162 (30), 91 (100), 78 (88), 77 (44), 57 (34). Anal. calc. for C₁₀H₁₀N₂O₂S₂ (254.33): C 47.23, H 3.96, N 11.01; found: C 47.04, H 3.93, N 11.13.

N-[(4-Methylphenyl)sulfonyl]hexanimidoyl Thiocyanate (**4d**). Yellow oil. Yield: 0.19 g (62%). IR (KBr): 2233, 1520, 1401, 1369, 1270, 1118. ¹H-NMR: 0.90 (t, ³J=6.9, Me); 1.39–1.45 (m, CH₂); 1.47–1.53 (m, CH₂); 1.91–1.93 (m, CH₂); 2.17 (t, ³J=7.0, CH₂); 2.42 (s, Me); 7.39 (d, ³J=8.0, 2 arom. H); 7.92 (d, ³J=8.0, 2 arom. H). ¹³C-NMR: 13.9 (Me); 18.5 (CH₂); 22.2 (CH₂); 30.9 (CH₂); 32.3 (Me); 38.4 (CH₂); 115.1 (CN); 127.5 (2 CH); 130.6 (2 CH); 142.1 (C); 147.2 (C); 171.2 (C). MS: 310 (3, M⁺), 238 (16), 215 (13), 155 (100), 91 (51), 77 (41), 57 (52), 43 (49). Anal. calc. for C₁₄H₁₈N₂O₂S₂ (310.43): C 54.17, H 5.84, N 9.02; found: C 54.49, H 5.91, N 9.12.

N-(Phenylsulfonyl)hexanimidoyl Thiocyanate (**4e**). Yellow oil. Yield: 0.17 g (57%). IR (KBr): 2246, 1517, 1408, 1348, 1268, 1141, 1086. ¹H-NMR: 0.91 (t, ³J=6.9, Me); 1.39–1.46 (m, CH₂); 1.48–1.53 (m, CH₂); 1.91–1.93 (m, CH₂); 2.18 (t, ³J=7.0, CH₂); 2.62 (t, ³J=8.0, 2 arom. H); 7.74 (t, ³J=8.0, 1 arom. H); 8.05 (d, ³J=8.0, 2 arom. H). ¹³C-NMR: 14.0 (Me); 18.5 (CH₂); 22.2 (CH₂); 30.9 (CH₂); 38.4 (CH₂); 115.2 (CN); 127.4 (2 CH); 130.1 (2 CH); 135.7 (CH); 147.4 (C); 170.9 (C). MS: 296 (1, M⁺), 270 (12), 224 (10), 155 (34), 141 (100), 77 (35), 71 (77), 43 (50). Anal. calc. for C₁₃H₁₆N₂O₂S₂ (296.41): C 52.68, H 5.44, N 9.45; found: C 53.00, H 5.52, N 9.57.

N-(Methylsulfonyl)hexanimidoyl Thiocyanate (**4f**). Yellow oil. Yield: 0.12 g (53%). IR (KBr): 2237, 1522, 1403, 1345, 1270, 1119, 1040. ¹H-NMR: 0.90 (t, ³J=6.9, Me); 1.38–1.45 (m, CH₂); 1.48–1.52 (m, CH₂); 1.91–1.94 (m, CH₂); 2.16 (t, ³J=7.0, CH₂); 3.56 (s, Me). ¹³C-NMR: 14.0 (Me); 18.5 (CH₂); 22.2 (CH₂); 30.9 (CH₂); 32.3 (Me); 39.4 (CH₂); 115.6 (CN); 170.7 (C). MS: 234 (1, M⁺), 208 (12), 176 (10), 155 (39), 78 (100), 71 (32), 57 (46), 43 (21). Anal. calc. for C₈H₁₄N₂O₂S₂ (234.05): C 41.00, H 6.02, N 11.95; found: C 41.36, H 6.09, N 12.06.

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